Enquiry for Course Details

	adam O	emicel Science (Coredite)	Appederation Ma	2022				
		emical Science (6 credits)	Academic Year					
Offering Department	Chemistry Quota 60							
Course Co-ordinator	Prof X D Li, Chemistry < xiangli@hku.hk >							
Teachers Involved	((Visiting Professor) Dr X H Yan,Chemistry) (Prof X D Li,Chemistry)							
Course Objectives	Modern chemistry is thought to be the "central science" as it plays a critical role in related biological, physical medical, and engineering disciplines. This course aims to introduce students to the newest concepts and technological breakthroughs in chemical sciences. Throughout the course, students will be introduced to how the interplay among molecules, materials, and interfaces leads to unprecedented functionalities that contribute to innovations in biology and medicine, smart materials, and sustainable energy schemes.							
Course Contents & Topics	Current topics focus on the interdisciplinary area of chemistry with biology, and material sciences. Covered topics include chemical genetics, epigenetics and proteomics; chemical biology for drug discovery and development stimuli-responsive nanomaterials; autonomous macromolecular motion; future power landscape; renewable energy conversion and utilization. Examples in protein posttranslational modifications, active colloidal thermoelectric materials, molecular machines, advanced rechargeable batteries, and next-generation fuel cells and electrolysers will be discussed.							
Course Learning Outcomes	On successful completion of this course, students should be able to:							
	CLO 1 understand important principles and topical trends in chemical sciences							
	CLO 2 demonstrate understanding of future directions in biomedical chemistry, nanomatertials, and energy sciences and applying this knowledge in comparing and contrasting various emergent technologies							
	CLO 3 interpret and analyse recent published research data in the field of chemistry and extract relevant chemical information to explain the observed properties and phenomena associated to the chemical systems							
Pre-requisites (and Co-requisites and Impermissible combinations)	Pass in CHEM3341 and CHEM3441.							
Course Status with Related Major/Minor /Professional Core	 2023 Major in Chemistry (Disciplinary Elective) 2023 Major in Chemistry (Intensive) (Disciplinary Elective) 2023 Minor in Chemistry (Disciplinary Elective) 2022 Major in Chemistry (Disciplinary Elective) 2022 Major in Chemistry (Intensive) (Disciplinary Elective) 2022 Major in Chemistry (Disciplinary Elective) 2021 Major in Chemistry (Disciplinary Elective) 2020 Major in Chemistry (Disciplinary Elective) 2020 Major in Chemistry (Disciplinary Elective) 2020 Major in Chemistry (Intensive) (Disciplinary Elective) 2020 Major in Chemistry (Intensive) (Disciplinary Elective) 2020 Major in Chemistry (Intensive) (Disciplinary Elective) 2020 Major in Chemistry (Intensive) (Disciplinary Elective) 2030 Major in Chemistry (Disciplinary Elective) 2040 Major in Chemistry (Disciplinary Elective) 2050 Major in Chemistry (Disciplinary Elective) 2019 Major in Chemistry (Disciplinary Elective) 2019 Minor in Chemistry (Disciplinary Elective) 2019 Minor in Chemistry (Disciplinary Elective) 							
Course to PLO Mapping	2023 Major in Chemistry < PLO 1,2,3,5 > 2023 Major in Chemistry (Intensive) < PLO 1,2,3,5 > 2022 Major in Chemistry < PLO 1,2,3,5 > 2022 Major in Chemistry (Intensive) < PLO 1,2,3,5 > 2021 Major in Chemistry < PLO 1,2,3,5 > 2021 Major in Chemistry (Intensive) < PLO 1,2,3,5 > 2020 Major in Chemistry < PLO 1,2,3,5 > 2020 Major in Chemistry (Intensive) < PLO 1,2,3,5 > 2020 Major in Chemistry (Intensive) < PLO 1,2,3,5 > 2019 Major in Chemistry < PLO 1,2,3,5 > 2019 Major in Chemistry (Intensive) < PLO 1,2,3,5 >							
Offer in 2023 - 2024	Y 2	nd sem	Examination	Мау				
Offer in 2024 - 2025	Y							
Course Grade	A+ to F							

		1							
Grade Descriptors	A Demonstrate thorough knowledge and understanding of essential facts, concepts and principles in chemical sciences, especially those relating to biomedical chemistry, nanomaterials, and energy sciences. Show strong ability to apply and integrate knowledge in chemical sciences in explaining the formation and properties of chemical species and in designing different chemical systems. Show strong ability to analyse and interpret experimental data to draw appropriate conclusions relating to the advanced principles and properties of chemical systems.								
	В	Demonstrate substantial knowledge and understanding of essential facts, concepts and principles in chemical sciences, especially those relating to biomedical chemistry, nanomaterials, and energy sciences. Show evidence to apply and integrate knowledge in chemical sciences in explaining the formation and properties of chemical species and in designing different chemical systems. Show evidence to analyse and interpret experimental data to draw appropriate conclusions relating to the advanced principles and properties of chemical systems.							
	С	Demonstrate general but incomplete amount of knowledge and understanding of essential facts, concepts and principles in chemical sciences, especially those relating to biomedical chemistry, nanomaterials, and energy sciences. Show some ability to apply and integrate knowledge in chemical sciences in explaining the formation and properties of chemical species and in designing different chemical systems. Show some ability to analyse and interpret experimental data to draw appropriate conclusions relating to the advanced principles and properties of chemical systems.							
	D	Demonstrate partial but incomplete command of knowledge and understanding of essential facts, concepts and principles in chemical sciences, especially those relating to biomedical chemistry, nanomaterials, and energy sciences. Show evidence of limited ability to apply and integrate knowledge in chemical sciences in explaining the formation and properties of chemical species and in designing different chemical systems. Show evidence of limited ability to analyse and interpret experimental data to draw appropriate conclusions relating to the advanced principles and properties of chemical systems.							
	Fail	Demonstrate little or no evidence of command of knowledge and understanding of essential facts, concepts and principles in chemical sciences, especially those relating to biomedical chemistry, nanomaterials, and energy sciences. Show little or no ability to apply and integrate knowledge in chemical sciences in explaining the formation and properties of chemical species and in designing different chemical systems. Show little or no ability to analyse and interpret experimental data to draw appropriate conclusions relating to the advanced principles and properties of chemical systems.							
Course Type	Lecture-b	Lecture-based course							
Course Teaching & Learning Activities	Activities			Details	Details No. of Ho				
	Lectures					36			
	Tutorials					12			
	Reading / Self study					100			
Assessment Methods and Weighting	Methods	5	Details		Weighting in final course grade (%)	Assessment Methods to CLO Mapping			
	Assignments		(20% Tests/Assignments; 5% participation)		25	CLO 1,2,3			
	Examina	ition			40	CLO 1,2,3			
	Presentation				35	CLO 1,2,3			
Required/recommended reading and online materials	References to specialized texts and other published materials will be made throughout the course.								
Course Website									
Additional Course Information	This course is also offered to RPg students, and the course code for RPg students is CHEM6118.								